

Supporting Information for

$\text{Bi}_x\text{Ti}_{1-x}\text{O}_z$ Functionalized Heterojunction Anode

with an Enhanced Reactive Chlorine Generation

Efficiency in Dilute Aqueous Solutions.

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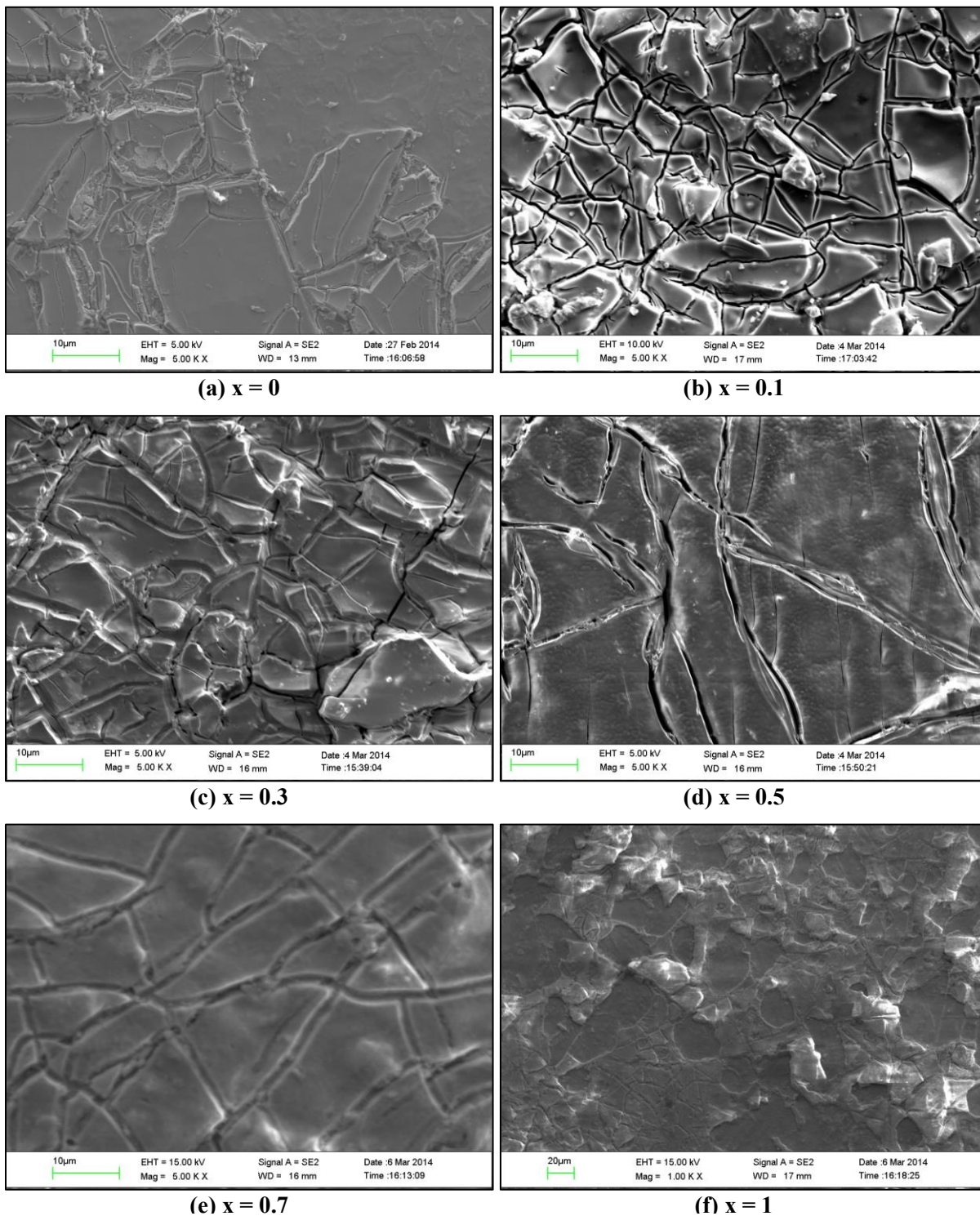
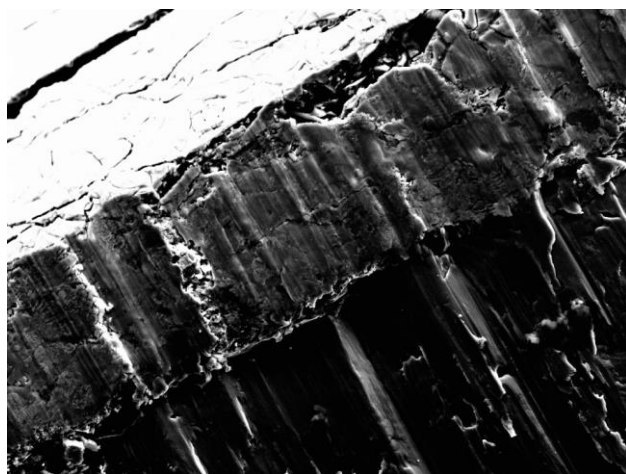
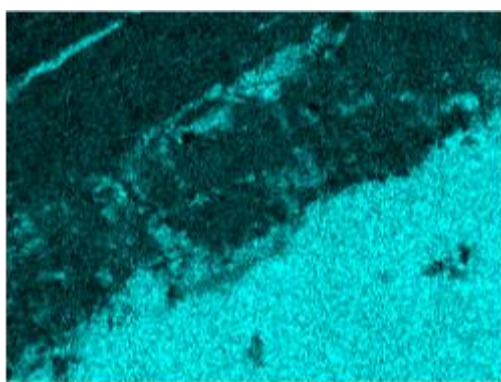


Figure S1. Scanning Electron Microscopy (SEM) images of horizontal view for $\text{Ir}_{0.7}\text{Ta}_{0.3}\text{O}_y/\text{Bi}_x\text{Ti}_{1-x}\text{O}_z$ hetero-junction anodes with variable molar fraction of Bi (x) with magnification of $\times 5000$ (a – e) or $\times 1000$ (f).

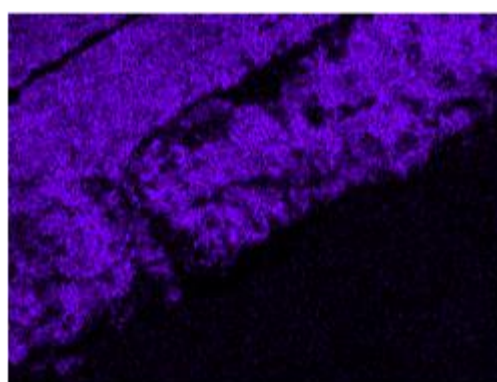


(a) SEM Image



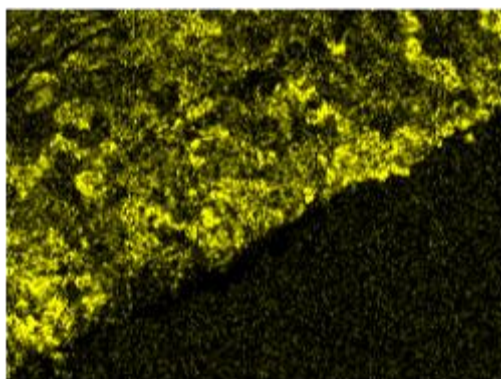
50μm

(b) Ti



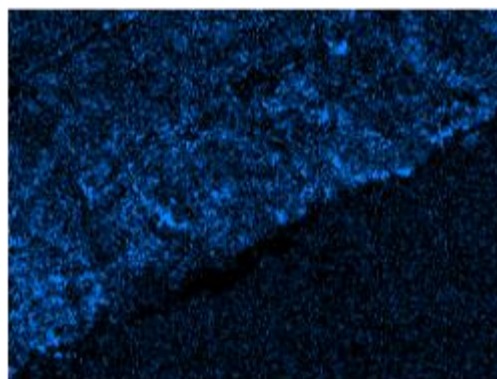
50μm

(c) Bi



50μm

(d) Ir



50μm

(e) Ta

Figure S2. (a) Scanning Electron Microscopy (SEM) image of cross-section view for $\text{Ir}_{0.7}\text{Ta}_{0.3}\text{O}_y$ / $\text{Bi}_{0.3}\text{Ti}_{0.7}\text{O}_z$ hetero-junction anode with Energy Dispersive Spectroscopy (EDS) mapping of (b) Ti, (c) Bi, (d) Ir, and (e) Ta.

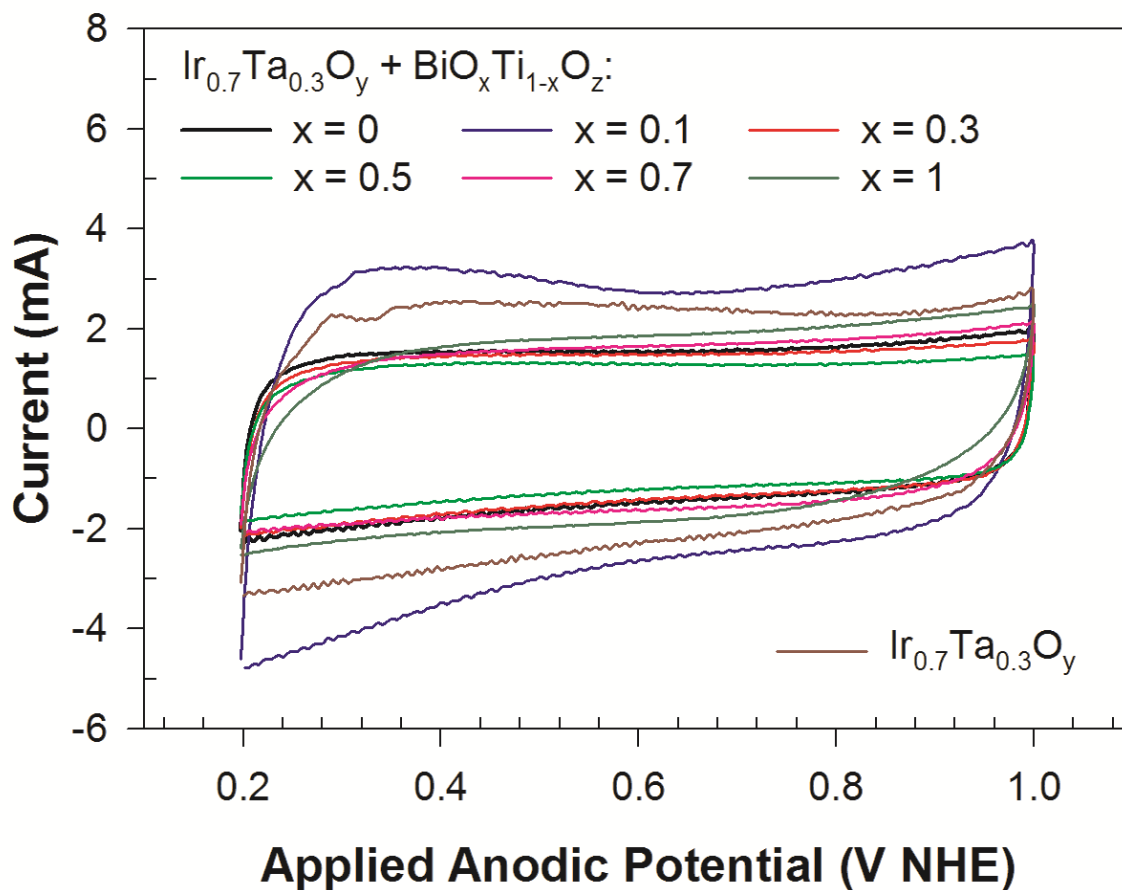


Figure S3. Cyclic voltammogram of $\text{Ir}_{0.7}\text{Ta}_{0.3}\text{O}_y/\text{BiO}_x\text{Ti}_{1-x}\text{O}_z$ hetero-junction anodes with variable molar fraction of Bi (x) for estimation of integrated charge (Q^*); counter electrode: stainless steel, geometric surface area: $3 \times 2 \text{ cm}^2$, electrolyte: 50 mM NaCl, scan range: 0.2 – 1.0 V, scan rate: 20 mV s^{-1} . Data for $\text{Ir}_{0.7}\text{Ta}_{0.3}\text{O}_y$ anode are shown as references.